FURTHER EVIDENCE FOR THE INVOLVEMENT OF A MONOCYCLIC β-LACTAM IN **THE ENZYMATIC CONVERSION OF 6-L-a-AMINOAbIPOYL-L-CYSTEINYL-D-VALINE INTO ISOPENICILLIN N.**

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Abstract: Incubation of δ - L - α -aminoadipoyl- L -[3-13C]cysteinyl- D -[3-²H]valine with Isopenicillin N Synthase (IPNS) resulted in the observation of a 'shunt metabolite', which we believe is formed from the collapse of an enzyme bound monocyclic β lactam intermediate. Chemical studies into the origin of the shunt metabolite suggest its formation occurred after initial β -lactam ring closure. Further chemical studies on the decomposition pathway of a free thiol monocyclic b-lactam have indicated it is not the source of the shunt metabolite, as upon decomposition the major product formed retains sulphur in the form of an ene-thiol dehydrocysteine.

Isopenicillin N Synthase (IPNS) is the enzyme which catalyses the cyclization of the Arnstein tripeptide δ -L- α -aminoadipoyl-L-cysteinyl-D-valine (LLD-ACV) (1) into isopenicillin N (2) ,^{1,2} This process has been shown to occur in a step wise fashion^{3,4} each closure exhibiting a marked primary kinetic isotope effect.5 Evidence to date strongly suggests that initial closure of the B-lactam ring occurs, followed by thiazolidine ring closure en route to isopenicillin N formation.⁶ (See Scheme 1, path a).

The stepwise nature of this reaction suggested to us that it may be possible to observe a monocyclic intermediate or other products formed as a result of leakage of the partially processed substrate from the enzyme before thiazolidine ring closure occurred (See Scheme 1, path b). To further increase the possibility of such

observations it was deemed adventitious to retard, by means of kinetic isotope effects, or even prevent thiazolidine ring closure by presenting IPNS with modified substrates. Thus two analogues were synthesised firstly by the incorporation of deuterium instead of hydrogen at the C-3 position of valine, e.g. (3a). and secondly the δ - L - α -aminoadipoyl- L -cysteinyl-glycine (5a) tripeptide which by its nature would be unable to participate in thiazolidine ring closure. An informative method for observation of a shunt pathway between $LLD-ACV$ (1) and Isopenicillin N (2) would be to follow the fate of the cysteinyl residue of ACV through the enzymatic process. Thus initially a series of specifically labelled tripeptides $LLD-A-[3-13C]C-[15N,$ $3-2H$]V (3b), LLD-A-[3-13C]C-[15N]V (3c), LLL-A-[3-13C]C-[15N, 3-2H]V (3d) were prepared from their benzyl protected precursors $(4b-d)^{7,8}$ (see scheme 2).

Scheme 2.

- **(i) EEDQ/NEts/and** appropiate valine ester, tosylate salt.
- (ii) EEDQ/NEts/glycine **benzyl ester, tosylate salt.**
- (iii) Tripeptide (4a): TFA/Anisole, Tripeptides (4b->4f, 6): Na/NH₃(1).
- **(iv) Hg(II). H2S. Oz/pH 8-9, reverse phase HPLC.**

Incubation of the tripeptides (3b and 3c) with IPNS under identical conditions⁹, followed by examination of their $C-13$ n.m.r. spectra in D_2O after protein precipitation and freeze drying gave the n.m.r. spectra shown (Figure 1). Control experiments with the tripepide (3d) and with denatured enzyme were also obtained for comparison. The major difference observed between the spectra was the presence of the resonances at δ c 89 in the case of (3b), which were approximately 1-5% of the intensity of the corresponding isopenicillin N product. This result was consistently achieved when using both native fungal and recombinant enzymes⁹, with several separately prepared samples of the tripeptide (3b). The nature of the signals at δ_c 89 (approximately 1:1 intensity, $\Delta \delta_C$ 7Hz) suggested that a ¹³C-¹⁵N coupling may be responsible for the δ_C 89 apparent doublet. This idea was dismissed after synthesis and incubation of the tripeptide δ -L- α -aminoadipoyl-L-[3-13C]cysteinyl-D-[3-2H]valine, (LLD-A-[3-13C]C-[3-²HJV, (3e)), [and (3f) as a control], as the splitting of the δ C 89 signals was still present in the absence of the $15N$ label. It became clear that the signals were not the components of a doublet, but instead probably represented a 1:l mixture of diastereotopic compounds.

In order to further investigate the nature of the δ_C 89 signals the tripeptide δ -L- α aminoadipoyl-L- $[3-13C]$ cysteinyl- $[15N]$ glycine (5b) was synthesised from (6) (see scheme 2). and incubated with IPNS under identical conditions again this incubation showed a signal at δ_C 89, although this time only as one singlet. As expected the major component in the n.m.r. spectrum (BB, C-13) was that of starting material. Literature searches¹⁰ revealed that the most likely moiety responsible for δ C 89 would be a hydrated aldehyde (i.e. -CH(OH)_2). In support of this idea the C-13 n.m.r. of incubation mixtures (CW mode) indicated that one hydrogen atom was indeed attached to each of the carbons giving rise to the δ 89 signals. Model studies were therefore undertaken in order to identify the molecular structure of the shunt metabolite. One such model was the aldehyde (7) derived from the ring opening of Penicillin G sulphone by propylamine, followed by treatment with HCl and extraction.¹¹ The "aldehyde (7) " actually exists in several forms (i.e. aldehyde/hydrate/enol) and could be observed to change (by C-13 n.m.r) from the aldehyde to the hydrate when water was added to a DMSO solution of the aldehyde (7). If D₂O was added the resonance assigned to \mathbf{C} H α in the C-13 n.m.r spectrum broadened and disappeared, suggesting exchange of the $CH\alpha$ proton with solvent. Furthermore, the H-1 n.m.r showed the hydrate resonance $-CH(OH)_2$, which appeared as a singlet when D20 was added to the DMSO solution of the aldehyde. When a DEPT experiment (D_2O) was carried out on the incubation mixtures (observing the signals at δ _C 89), no protons were observed on the carbon atom α to the carbons responsible for the δ C 89 signals, in agreement with the findings of the work on the model compounds. The exact structure of the 'shunt metabolite' responsible for the δ \sim 89 signals observed from (3e) was therefore tentatively assigned as being that shown in structure (8c).

A totally synthetic sample of this material was prepared by carrying out an ozonolysis (MeOH/H₂O, -50 $^{\circ}$ C), followed by a DMS work up of the tripeptide (9a), which was prepared by standard methodology from the acid labile protected tripeptide (10a) (scheme 3). Similarly (9b) was converted to (8b).

Scheme 3.

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(1) isobutylchloroformate/NEt<sub>3</sub>/L-vinylglycine, (ii) EEDQ/NEt<sub>3</sub>/D-valine benzhydryl ester, tosylate salt.
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Access to an authentic sample (8a) of the shunt metabolite enabled a thorough spectroscopic comparison to be carried out, all the data obtained for the authentic material was in complete agreement with that obtained from the enzymatically derived shunt compound. Noticably the hydrated aldehydes (8a,8b) obtained via ozonolysis of the tripeptides (9a, and 9b), gave the signals at δ_C 89 (BB) as two and one resonance respectively, in agreement with the enzyme derived shunt metabolitesfrom (3e) and (5b).

HPLC purification of a crude incubation mixture of $LLD-A[3-13C]C-[3-2H]V$ (3e) with IPNS (ODS, mobile phase 20mM NH₄HCO₃), gave a fraction with the same retention time as a synthetic sample of the hydrate which contained the resonances at δ C 89. This sample also gave a H-l n.m.r. which showed signals that were identified as the

⁽iii) EEDQ/NEt₃/glycine benzhydryl ester, tosylate salt, (iv) TFA/Anisole, (v) O₃, MeOH/H₂O,-50°C/DMS.

hydrate ${}^{13}CH(OH)_2$. Treatment of this hydrate (obtained both synthetically and enzymatically) with sodium borohydride gave two new signals with equal intesities at δ _C 62 which were identical to the signals obtained for the -CH₂OH carbons from the two tripeptides δ -L- α -aminoadipoyl-L-serinyl-D-valine (13a) and δ -L- α aminoadipoyl-D-serinyl-Q-valine (13b) (These were prepared from the fully benzyl protected tripeptides (14a,14b) by hydrogenation, see scheme 4).

(i) Hz/lo% Pd/C/THF/H20, (ii) **Reverse** phase HPLC.

In order to further establish the identical nature of the synthetically and enzymatically derived hydrate, a second HPLC system was employed $(\mu$ -bondpack-NH2, mobile phase 25% MeOH/ 75% (0.05% aq HCOOH)), here again both samples were shown to have identical retention times. Isolation of the "hydrate window" from the HPLC of a crude incubation mixture of δ - L - α -aminoadipoyl- L -cysteinyl-[3- $2H$]- D valine (3a) allowed a ¹H n.m.r. of the partially purified shunt metabolite (8d) to be obtained. When the sample was doped with the synthetic material (8a) the resonances associated with the hydrate proton $-CH(OH)_2$ were enhanced in intensity as anticipated (see figure 2).

Isopcnicillin N ⁴⁶³

In order to determine whether the shunt metaholite was derived from an enzyme free mercaptoazetidinone (15b). a synthetic sample of the unstable thiol (15b) was prepared¹² by liberating it from its stable mercury(II) salt (15a), following treatment with H2S. The mercaptoazetidinone was then exposed to conditions identical to those encountered in IPNS incubations. Upon investigation of the total decomposition products, the C-13 n.m.r. spectrum showed no signs of a δ C 89 signal, while the proton n.m.r. showed that the major decomposition product gave rise to a singlet at δ _H 8.11. Unsuccessful attempts to directly isolate the decomposition product giving rise to the δ H 8.11 signal, was followed by the preparation of an authentic sample of the suspected decomposition product, the thiazole (16b) (see scheme 5).

Having the relevant spectroscopic and HPLC data it became clear that the decomposition product was not the thiazole, and on this basis the ene-thiol (17) was considered. Thus iodoacetic acid was added to the crude decomposition mixture in order to trap out the putative ene-thiol species. This approach proved successful and allowed HPLC isolation and characterization of the alkylated dehydrocysteine containing tripeptide (18). as a major decomposition product of mercaptoazetidinone (15b).

A possible route whereby the monocyclic enzyme-bound β -lactam intermediate gives rise to the shunt metabolite and yet completes the stoichiometric reduction of dioxygen to water in each catalytic cycle is shown in scheme 6.

As already discussed, the deuterium isotope effect competitively slows carbonsulphur bond formation to Isopenicillin N (path a), thus promoting fragmentation

464 **J. E. BALDWIN et al.**

(path b) to the imminium ion (21), atomic sulphur and iron(I1). Quenching this imminium ion gives rise to the hydroxy- β -lactam.^{*} It has already been shown that such 4-hydroxyazetidinones are unstable with respect to ring opening, 15 forming aldehydes and their derived enols analogous to those we have observed as the shunt metabolite (8).

More recently¹⁶ we have shown in our laboratory that incubation of the tripeptide δ - $L-\alpha$ -aminoadipoyl-L-homocysteinyl-D-valine (20) with IPNS gave rise to analogous hydroxylated y-lactams, this result is readily interpreted on the basis of scheme 6.

In conclusion we have observed the formation of a shunt type metabolite from IPNS as it transforms its natural substrate into isopenicillin N. The nature of the shunt metabolite has been assigned to that of the hydrated aldehyde (8), which we believe is derived from the release of a hydroxy- β -lactam (19) formed as a result of a diversion of the IPNS enzyme from its normal catalytic pathway.

Such a species has also recently been proposed in the biosynthesis of Clavulanic acid.¹⁴

Experimental.

Machine details.

¹H nuclear magnetic resonance spectra were recorded on Bruker AM 500 (500MHz), Bruker WH 300 (3OOMHz), Bruker AM 250 (25OMHz) or Varian Gemini 200 (2OOMHz) spectrometers. 13 C nuclear magnetic resonance spectra were recorded on Bruker AM 500 (126MHz) or Varian Gemini 200 (SOMHz), multiplicities were assigned by off resonance proton decoupling or DEPT spectrum editing. [All chemical shifts (6) are expressed in parts per million (ppm)]. 1 H spectra were referenced internally to residual CHCl₃ (δ _H 7.27) and CHD₂SOCD₃ (δ _H 2.50) in CDCl₃ or CD₃SOCD₃ respectively, while aqueous solutions (D₂O) were referenced internally to TSP (δ H 0.00). ¹³C spectra were referenced internally to CDCl₃ (δ C 77.00), dioxan for aqueous solutions (δ C 67.00) and CD_3 SO CD_3 (δ_C 39.70). Unless otherwise stated the pD of aqueous solutions was 6-7, and were HOD suppressed. High performance (pressure) liquid chromatography (HPLC) was performed on one of two systems depending on sample quantity: (1) . Preparative scale. $(>5mg)$.

Two Gilson model 303 pumps, Rheodyne 7125 injector (lOOOp1 loop), Gilson model HM holochrome detector (set at 214nm), and a 250x10.0mm i.d. column packed with Zorbax[®] hypersil ODS (particle size 5μ m), with a pre-column (30x10.0mm i.d. packed with hypersil ODS). Flow rates were typically 3-4ml/min, with computer controlled gradient operation.

(2) . Semi-analytical scale. $(\langle 1mg \rangle)$.

Two Waters model 510 pumps, Rheodyne 7125 injector (200µl loop), Waters model 441 and model 440 absorbance detectors (in series, $\lambda = 214$ and 254nm respectively), with a $250x4.6mm$ i.d. column packed with Zorbax® hypersil ODS (particle size $5\mu m$), and a pre-column (lSx4.6mm i.d. packed with hypersil ODS). Flow rates were typically lml/min, and were controlled with a Waters automated gradient controller. All solvents and samples were filtered $(0.5\mu m)$ and rigorously degassed before use. Mass spectrometry was performed on a V.G. Mass Lab. 20-250 Quadrupole (ACE (alternative electron impact/chemical ionization)), FAB (+ve argon fast atom bombardment), a V.G. Micromass ZAB-IF, DC1 (desorption chemical ionization), FAB (+ve argon) spectrometers. Optical rotations were run on a Perkin-Elmer 241 polarimeter. Micro analysis were performed on a Carlo-Erba Strumentazone Elemental analyser, model 1106. Flash chromatography was performed using Merck silica gel 60. Analytical TLC was performed on commercial Merck silica gel 60 F₂₅₄ aluminium backed plates (0.2mm thickness). Where appropriate reagents and solvents were purified by standard methods.

 $DL-[3-13C]$ -S-benzyl cysteine was prepared from ¹³C paraformaldehyde (90 atom %, Amersham) obtained from an aqueous solution of $13C$ formaldehyde (9.1% w/v) using, with only slight variation, the method of Upson et al.¹⁷ The S-benzyl-DL- $[3-13C]$ cysteine was resolved via N-acetylated S-benzyl-DL- $[3-13C]$ -cysteine using hog kidney acylase to give, after ion exchange chromatography, S-benzyl- L -[3.13C]cysteine.¹⁸ DL-[3-2H]valine and DL-[15N,3-2H]valine were prepared by the method of

J. E. BALDWIN et *al.*

Baldwin et al.¹⁹ and where necessary were resolved using the N-acetyl derivative and hog kidney acylase¹⁸. After treatment with this enzyme the various N-acetyl Dvalines were extracted into EtOAc and crystallised $(x2)$. The amino acids D-valine, DL- $[15N,3-2H]$ valine, D- $[14N,3-2H]$ valine and $[15N]$ glycine were all converted to their benzyl ester ammonium tosylate salts by treatment with benzyl alcohol and TsOH according to usual literature procedures²⁰. Additionally D -valine, D -[3-²H]valine and glycine were prepared as their benzhydryl ester ammonium tosylate salts using diphenyldiazomethane²¹. S-4-methoxybenzyl-L-cysteine was prepared by the literature method²². L-vinylglycine was prepared from L-methionine in five steps by the method of Rapoport et $a^{1,23}$ in an overall yield of 23% with no observable racemisation. $LLD-ACV$ is a literature compound⁷ and was used as a reference for comparison.

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(A). EEDO Couplings of Dipeptides to Protected Amino acids.⁷

Equimolar quantities of EEDQ, NEt_3 , benzyl or benzhydryl protected amino acid tosylate salts, and the free carboxylic acid dipeptides (typically 0.3mmol) were dissolved in dry DCM (5ml), with a small quantity of $Na₂SO₄$ and strirred overnight under $Ar(g)$. EtOAc (30ml) was added, and the solution washed sequentially with NaHCO₃ (saturated solution, 30ml), HCl $(1M, 30ml)$, and NaCl (saturated solution, 30ml), dried (Na₂SO₄), filtered, and the solvent removed in vacuo, to give the crude fully protected tripeptide. These were purified by column chromatography on silica gel as described for the individual tripeptides.

(B) . Acid deprotections of Tripeptides.

The fully protected tripeptide (typically lOOmg, O.lmmol), was dissolved in freshly distilled TFA (2ml) and dry anisole (400 μ l), and the mixture refluxed under Ar(g) for 15min. After cooling the TFA was removed by azeotroping with toluene $(3x5ml)$, and the residue was partitioned between water (10ml) and EtOAc (10ml). The aqueous layer was washed with further EtOAc (lOml), before freeze drying, to give the tripeptide as its TFA salt. Treatment was then as described for each of the individual tripeptides.

$f(C)$ Na/NH₃(1) deprotections of Benzyl protected Tripeptides.⁷

The benzyl protected tripeptides (typically 50mg, O.OSmmol) were dissolved in freshly distilled THF (2ml), to which sodium dried liquid ammonia (30ml) was added by distillation, at -35cC. Small pieces of cleanly cut sodium metal were added to the mixture at -35W until the blue colour persisted for 10min. At this time the reaction was quenched by the addition of a small quantity of $NH₄OAC$, with rapid stirring. The ammonia was removed under a stream of $Ar(g)$, at room temperature, and the remaining solvent was removed under high vacuum. This residue was dissolved in 10% AcOH (20ml), washed with EtOAc ($2x20$ ml), before the dropwise addition of 5% $Hg(OAc)_2$ in 10% AcOH. The resulting white precipitate was collected by centrifugation, washed with water (2x20ml), MeOH (20ml), and water (20ml), suspended in water

(2ml), and the free thiol liberated by passing $H_2S(g)$ through the solution for 5 min. The solution was centrifuged, before filtration through washed celite, and freeze drying. The residue was taken up in water (10ml), the pH adjusted to 9-10, and O_2 bubbled through the solution for 4h. This solution was freeze-dried, before the residue was purified by reverse phase HPLC (stationary phase ODS, mobile phase 25mM NH4HCO3/MeOH).

(D) . Dipeptide Formation with Isobutylchloroformate.⁷

 $N-Benzyloxycarbonyl-\alpha-benzylester-\delta-L-\alpha-aminoadipic acid, or N-4-$

 $methoxybenzyloxycarbonyl-\alpha-4-methoxybenzylester-\delta-L-\alpha-aminoadipic acid$ (typically O.Smmol) was dissolved in freshly distilled THF (6ml). together with NEt₃(leq), and cooled to -15°C for 15min. Isobutyl chloroformate (leq) was added and the mixture stirred at -15° C for 30min. To this mixture the amino acid (1eq, Lvinylglycine, or S-protected- L -cysteine) in water (6ml) and NEt₃ (2eq), cooled to $0^{\circ}C$, was added rapidly in one portion. The mixture was vigorously stirred at room temperature for 1h. Water (10ml) was added, and the solution washed with $Et₂O$ (2x20ml), followed by acidification to pH 1-2 (1 \underline{M} HCl), and the white precipitate extracted with EtOAc (3x50ml). The organic layer was washed with NaCl (saturated solution, 30ml), dried (Na₂SO₄), filtered, and the solvent removed in vacuo, to give the crude dipeptide as a foam or oil. These were purified by crystallization where appropiate.

(E) . Incubation Conditions of IPNS with Substrates.

IPNS was typically available in Tris buffer, which was exchanged with $50 \text{m} \text{M}$ NH₄HCO₃ before use by the passage through an equilibrated Sephadex G-25 gel filtration column. IPNS, which typically had an activity of 5 I.U./ml, was used with a substrate ratio of approximately lmg/ml of IPNS solution. The substrate (lmg) was dissolved in 50mM NH₄HCO₃ (3ml), to which was added sequentially DTT (100 μ l, 100mM), ascorbic acid (100 μ l, 50mM), and Iron (II) sulphate (100 μ l, 5mM), followed by the enzyme solution. The mixture was divided into two, to aid aeration, and shaken at 250 r.p.m. at 27°C for 10min. Additional DTT (100 μ 1, 100mM) and Iron(II)sulphate (100 μ 1, 5mM) were added, and the mixture shaken for a further IOmin, after which time the enzyme was precipitated by the addition of acetone (20ml), the mixture centrifuged (10,000g, 4°C), and the clear supernatant decanted. The solvent was removed in vacuo, and the residue freeze-dried, to give the "crude incubation mixture".

N-Benzyloxycarbonyl-a-benzylester-δ-L-a-aminoadipoyl-S-benzyl-L-[3-
¹³Clcvsteinvl-[¹⁵Nlglvcine.benzylester (6).

 15 N-Glycine benzylester (0.035g, 0.22mmol), was coupled to benzyl protected LL -'A[3-¹³C]' (leq), using IIDQ (leq, 0.082g), in DMF (5ml) with slight modification of general procedure A.7 Purification by column chromatography on silica gel [eluant EtOAc/Petrol (2:l) Rf 0.41, gave the title compound, which was recrystallized from

EtOAc/Petrol to give a white solid $(0.09g, 56%)$; δ_H $(500MHz, CDCl₃)$, 7.38-7.22 (20H, complex m, ArH), 6.82 (1H, 2xt, J 93, 5Hz, CH₂¹⁵NH), 6.24 (1H, d, J 7Hz, NH), 5.58 (1H, d, J **8Hz. NH), 5.20-5.08 (6H,** m, 3xABq, 3xC&Ar), 4.54-4.49 and 4.44-4.38 (2H, 2xm. $2xCH\alpha$), 4.10-4.06 and 4.00-3.95 (2H, AB part of ABX system, J 18, 6, 5Hz, CH₂¹⁵NH), 3.74 (2H, **d,** J 6Hz, 13CSC&Ar), 3.04-3.00, 2.87-2.83, 2.76-2.71 and 2.59-2.55 (2H, AB part of ABMX system, J 143, 141.5, 14, 7, 6Hz, $^{13}CH_2S$), 2.26-2.12 (2H, 2xm, CH₂CO), 1.93-1.80 and 1.77-1.66 (4H, 2xm, CH₂CH₂CH₂CO); δ _C (126MHz, CDCl₃), 172.54, 172.05, 170.60 and 169.28 (4x& 2x amides, 2x esters), 156.09 (s, urethane), 138.04, 136.23, 135.08 and 135.04 (4xs, ArC-1), 128.99-127.28 (d, ArCH's), 67.21, 67.16 and 67.04 (3xt, 3xCH₂Ar), 53.6 (d, CH α of aminoadipoyl), 51.97 (d of d, J 48Hz, CH α -¹³C), 41.40 (d of t, J 17Hz, $15NH-CH_2$), 36.58 (t, SCH₂), 35.26 (t, CH₂CO), 33.11 (t, $13CH_2$), 31.91 (t, $CH_2CH_2CH_2CO$), 21.25 (t, $CH_2CH_2CH_2$); m/z (+ve argon FAB), MH+ 728.

<u>N·Benzyloxycarbonyl-α-benzylester-δ-L-α-aminoadipoyl-S-benzyl-L-13-
¹³Clcvsteinyl-D-13-²H lvaline, benzylester (4e),</u>

 D -[3-2H]Valine benzyl ester, tosylate salt (0.045g, 0.11 mmol), was coupled with 'LL- $A[3-13C]C⁷$ according to general method A. Purification by column chromatography on silica gel [eluant DCM/EtOAc $(3:1)$ Rf 0.5] gave the title compound as a white powder $(0.052g, 58\%)$; δ_H (500MHz, CDCl₃), 7.36-7.21 (20H, complex m, ArH), 6.75 (1H, d, J 8.5Hz. NH), 6.24 (lH, d, J 7Hz, NH), 5.58 (IH, d, J 8.5Hz, NH), 5.21-5.06 (6H, m, 3xABq, $3xCH₂Ar$, 4.53-4.49 (2H, 1xd and 1xm, J 9Hz, CH_{α} of valinyl and cysteinyl), 4.41-4.38 (1H, m, CH α of aminoadipoyl), 3.76 (2H, d, J 4Hz, SCH₂Ar), 3.01-2.97, 2.83-2.79, 2.72-2.68 and 2.55-2.51 (2H, AB part of ABMX system, J 142, 141, 14, 7, 6Hz, $^{13}CH_2S$), 2.20-2.09 (2H, 2xm, CH₂CO), 1.92-1.86 and 1.73-1.64 (4H, 2xm, CH₂CH₂CH₂CO), 0.89 and 0.83 (6H, 2xs, Me₂CD); m/z (+ve argon FAB), MH⁺ 770; m.p. 109-112°C; ^{[α]_n²⁰ -18°} $(c = 1, CHCl₃)$.

N -Benzvloxvcarbonvl- α -benzvlester- δ -L- α -aminoadipovl-S-benzvl-L- 3 -13Clcvsteinvl-D-[¹⁵N.3-²H lvaline, benzylester (4b) and N-. . **carbonvl** _ **a** - **benwlester** _- **6 L** _ *a* _ **amlnoadlDovl S benzvl** L r3 . - _ _ - **13Clcvsteinvl-L-**[15N.3-2H |valine, benzvlester (4d),

The diastereoisomeric mixture was prepared according to general method A, using benzyl protected $LL-A[3-13C]^7$ and $DL-[15N,3-2H]$ valine benzyl ester, tosylate salt (O.O71g, 0.17mmol). The resulting mixture of diastereoisomers was separated by column chromatography on silica gel [eluant EtOAc/Petrol (1:l) Rf 0.5 and 0.451, to give as the less polar component benzyl protected $LLD-A[3-13C]C[^{15}N,3-2H]V$ (4b), (0.052g, 40%); δ_H (500MHz, CDCl₃), 7.41-7.18 (20H, complex m, ArH), 6.76 (1H, 2xd, $2J_{NH}$ 92Hz, J 8.5Hz, $15NH$), 6.26 (1H, d, J 7Hz, NH), 5.59 (1H, d, J 8Hz, NH), 5.19-5.04 (6H, m, 3xABq, 3xCH₂Ar), 4.51 (1H, d, J 9Hz, CH α of valinyl), 4.45 (1H, m, CH α of aminoadipoyl), 3.78 (2H, d, J 4Hz, $SCH₂Ar$), 3.02-2.98, 2.82-2.78, 2.73-2.69 and 2.54-2.49 (2H, AB part of ABMX system, J 142, 141, 14, 7, 6Hz, 13 C $H₂$ S), 2.26-2.09 (2H,

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2xm, CH₂CO), 1.92-1.86 and 1.73-1.64 (4H, 2xm, CH₂CH₂CH₂CO), 0.89 and 0.84 (6H, 2xs, Me₂CD); m/z (+ve argon FAB), MH⁺ 771; m.p. 107-110°C; $[\alpha]_D^{20} -18$ ° (C = 1, CHCl₃). Benzyl protected LLL-A $[3-13C]C[15N,3-2H]V$ (4d) (more polar component), (0.050g, 38%); δ_H (500MHz, CDCl₃), 7.41-7.18 (20H, complex m, ArH), 6.86 (1H, 2xd, ²J_{NH} 90Hz, J 9Hz, 'SNH), 6.32 (lH, d, J 7Hz. NH), 5.57 (lH, d, J 8Hz, NH), 5.19-5.04 (6H, m, 3xABq. $3 \times CH_2Ar$, 4.53 (1H, d, J 9Hz, CH α of valinyl), 4.41 (1H, m, CH α of aminoadipoyl), 3.78 $(2H, d, J, 4Hz, SCH₂Ar), 3.03-2.98, 2.85-2.81, 2.74-2.70$ and $2.58-2.52$ $(2H, AB)$ part of ABMX system, J 142, 141, 14, 7, 6Hz, $^{13}CH_2S$), 2.23-2.09 (2H, 2xm, CH₂CO), 1.92-1.86 and 1.73-1.64 (4H, 2xm, CH₂CH₂CH₂CO), 0.89 and 0.83 (6H, 2xs, Me₂CD); m/z (+ve argon FAB), MH+ 771.

<u>N-4-Methoxybenzyloxycarbonyl-α-4-methoxybenzylester-δ-L-α-</u> aminoadipovl-S-4-methoxybenzyl-L-cysteinyl-D-(3-2H)valine, benzhydrylester (4a).

 $N-4-Methoxybenzyloxycarbonyl-\alpha-4-methoxybenzyl-\delta-L-\alpha-aminoadipoyl-S-4$ methoxybenzyl-L-cysteine (0.21g, 0.3mmol), was coupled to $D-(3-2H)$ valine benzhydryl ester (l.leq, O.lg) with EEDQ (O.O76g, leq) according to general method A, to give after purification by column chromatography on silica gel [eluant EtOAc/Petrol $(1:1)$], the title compound, $(0.21g, 77\%)$; δ_H (500MHz, CDCl₃), 7.32-7.24 and 6.89-6.77 (24H, 2xm, NH, CHPh2 and ArH), 6.29 (lH, d, J 7Hz, NH), 5.50 (lH, d, J 12Hz, NH), 5.16- 5.03 (4H, 2xABq, 2xCH₂Ar), 4.64 (1H, d, J 12Hz, CH α of valine), 4.57-4.51 and 4.38-4.36 (2H, 2xm, 2xCHa), 3.79 and 3.76 (9H, 2xs, 3xMeO), 3.72 (2H, s, SCH₂Ar), 2.89-2.79 and 2.70-2.60 (2H, AB part of ABX system), $2.15 - 2.09$ (2H, $2xm$, CH $2CO$), $1.92 - 1.82$ and 1.79-1.65 (4H, $2xm$, $CH_2CH_2CH_2CO$), 0.88 and 0.76 (6H, $2xs$, Me₂CD); δ_C (50MHz, CDCl3), 172.38, 170.38, 170.86 and 170.47 (4xs, 2x ester, 2x amides), 160.04, 159.78 and 159.03 (3xs, $3xArC-4$), 156.44 (s, 1x urethane), 140.06 and 139.93 (2xs, $PhC-1$), 130.34-127.07 (d's, ArcH's), 113.98-114.17 (3xd, 3xArC-3), 77.98 (d, CHPhz), 67.01 and 66.83 (2xt, 2xOCH₂Ar), 55.18 (q, 3xMeO), 57.18, 53.65 and 52.06 (3xd, 3xCH α), 35.71 (t, CHCH₂S), 35.20 (t, CH₂CO), 33.09 (t, SCH₂Ar), 31.67 (t, CHCH₂CH₂), 21.07 (t, $CH_2CH_2CH_2$), 18.85 and 17.10 (2xq, 2xMe₂CD); m.p. 45-50°C; m/z (+ve argon FAB), MNa⁺ 957; C₅₂H₅₈N₃O₁₁SD requires C 66.79, H 6.36, N, 4.49, S 3.43% : found C 66.81, H 6.61, N 4.39, S 3.29%; m.p. 45-50°C; $[\alpha]_0^{20}$ -0.2° (C = 1.07, CH₂Cl₂).

δ -(L- α -Aminoadipoyl)-L-cysteinyl-D-(3-²H)valine disulphide (3a).

 $(N-4-Methoxybenzyloxycarbonyl)-\alpha-4-methoxybenzylester-\delta-L-\alpha-aminoadipoyl-S-4$ methoxybenzyl- L -cysteinyl- D -(3-²H)valine, benzhydryl ester (4a) (0.089g, 0.1mmol), was deprotected, with TFA (2ml) and anisole $(200\mu l)$ according to general method B. The residue was taken up in water (10ml), the pH adjusted to 9-10, and O_2 bubbled through the solution for 2h. This crude disulphide was purified by reverse phase HPLC (Stationary Phase ODS, mobile phase 25mM NH₄HCO₃/20%MeOH) to give the title compound as a white powder (29mg, 89%); δ_H (500MHz, D₂O, pH 6), 4.06 (1H, s, CH α of valine), 3.71 (1H, dd J 2x6Hz, $CH\alpha$ of aminoadipoyl), 3.21-3.17 and 2.99-2.96 (2H, AB part of ABX system, J 14, 9, 5Hz, C H_2S), 2.39 (2H, dd, J 2x7Hz, C H_2CO), 1.90-1.83 and

1.81-1.64 (4H, 2xm, CH₂CH₂CH₂CO), 0.89 and 0.84 (6H, 2xs, Me₂CD), CH α of cysteine under HOD suppression; δ C (126MHz, D₂O), 178.42, 176.26, 174.79 and 171.63 (4xs, 2x acids, $2x$ amides), 61.16, 54.87 and 53.13 (3xd, $3xCH\alpha$), 39.17 (t, CH_2S), 35.20 (t, CH_2CO), 30.23 (t, CH₂CH₂CH₂CO), 21.29 (t, CH₂CH₂CH₂), 19.36 and 17.58 (2xq, 2xMe₂CD); m/z (+ve argon FAB), MH+ (disulphide) 727, MH+ (thiol) 365.

δ -(L- α -Aminoadipovl)-L-(3.¹³C)cysteinvl-D-(¹⁵N.3.²H)valine disulphide $(3b).7$

 $(N-Benzyloxycarbonyl)-\alpha-benzylester-\delta-({L-\alpha-Aminoadipoyl})-S-benzyl-L-(3- $-\alpha$)-S- $-\alpha$$ $13C)$ cysteinyl- $D^{-(15)}N,3^{-2}H$)valine benzylester (4b) (0.052g, 0.07mmol) was deprotected according to general method $C₁$ to give, after oxidation to the disulphide and purification by reverse phase HPLC (mobile phase 20m M NH₄HCO₃/10% MeOH) the title compound as a white powder (16mg, 63%); δ_H (500MHz, D₂O), 4.11 (1H, s, CH α of valine), 3.75 (1H, dd J 2x6Hz, CH α of aminoadipoyl), 3.38-2.85 (2H, AB part of ABMX system, J 142, 142, 14, 7, 6Hz, CH¹³CH₂S), 2.39 (2H, dd, J 2x7Hz, CH₂CO), 1.94-1.80 and 1.78-1.62 (4H, 2xm, CH_2CH_2CHO), 0.90 and 0.87 (6H, 2xs, Me_2CD); m/z (+ve argon FAB), MH+ (disulphide) 731, MH+ (thiol) 367.

 $-L-\alpha$ -Aminoadipovl) $-L$ - $(3$ - $^{13}C)$ cvsteinvl \cdot D- (^{15}N) valine disulphide (3c)

 $(N-Benzyloxycarbonyl)-\alpha-benzylester- δ -({ $L-\alpha$ -Aminoadipoyl)-S-benzyl- $L-(3-$$ $13C)$ cysteinyl-D- $(15N)$ valine benzylester $(4c)$ ⁷ $(0.072g, 0.09mmol)$, was deprotected according to general method C, with subsequent oxidation to the disulphide, purification being accomplished by reverse phase HPLC (mobile phase 20mM $NH_4HCO₃/20%MeOH$, to give the title compound as a white powder (26mg, 76%); δ_H (500MHz, D₂O), 4.07 (1H, s, CH α of valine), 3.72 (1H, dd, J 2x6Hz, CH α of aminoadipoyl), 3.37-3.33, 3.20-3.04 and 2.86-2.82 (2H. AB part of ABMX system, J 143, 142, 14, 9, 5Hz, CH¹³CH₂S), 2.40 (2H, dd, J 2x7Hz, CH₂CO), 2.13-2.06 (1H, m, CHMe₂), 1.93-1.82 and 1.76-1.66 (4H, 2xm, $CH_2CH_2CH_2CO$), 0.90 and 0.87 (6H, 2xd, J 7Hz, Me_2CH); m/z (+ve argon FAB), MH+ (disulphide) 729, MH+ (thiol) 366.

δ -(L- α -Aminoadipoyl)-L-(3-¹³C)cysteinyl-L-(¹⁵N,3-²H)valine disu $(3d)$.

 $(N-Benzylovycarbonyl)-\alpha-benzylester-\delta-(L-\alpha-Aminoadipoyl)-S-benzyl-L-(3 13C$)cysteinyl-L- $(15N, 3-2H)$ valine benzylester (4d) $(0.050g, 0.07mmol)$ was deprotected according to method C to give, after oxidation to the disulphide and purification by reverse phase HPLC (mobile phase 50mM NH₄HCO₃/25% MeOH), the title compound as a white powder (17mg, 72%); δ_H (500MHz, D₂O), 4.09 (1H, s, CH α of valine), $3.57-3.54$ (1H, m, $CH\alpha$ of aminoadipoyl), $3.46-2.81$ (2H, AB part of ABMX system, J 142, 141, 14, 7, 6Hz, CH¹³CH₂S), 2.39 (2H, dd, J 2x7Hz, CH₂CO), 1.88-1.63 (4H, 2xm, CH₂CH₂CH₂CO), 0.88 and 0.86 (6H, 2xs, Me₂CD); m/z (+ve argon FAB), MH⁺ (disulphide) 731.

 δ -(L- α -Aminoadipovl)-L- $(3.13C)$ cvsteinvl-D- $(3.2H)$ valine disulphide (3e).

 $(N-Benzyloxycarbonyl)-\alpha-benzylester-\delta-(L_{\lambda}^{K}Aminoadipoyl)-S-benzyl-L-(3 13C$)Cysteinyl- $D-(3-2H)$ Valine benzylester (4e) (0.052g, 0.07mmol), was deprotected according to general method C, followed by oxidation to the disulphide, and purification being accomplished by reverse phase HPLC (mobile phase 2OmH $NH_4HCO₃/20%MeOH$, to give the title compound as a white powder (19mg, 76%); δ_H (500MHz, D₂O), 4.07 (1H, s, CH α of valine), 3.72 (1H, dd, J 2x6Hz, CH α of aminoadipoyl), 3.37-3.33, 3.20-3.04 and 2.86-2.82 (2H, AB part of ABMX system, J 143, 142, 14, 9, 5Hz, CH¹³CH₂S), 2.40 (2H, dd, J 2x7Hz, CH₂CO), 1.93-1.82 and 1.76-1.66 (4H, 2xm, $CH_2CH_2CH_2CO$, 0.90 and 0.87 (6H, 2xs, Me₂CD); δ_C (126MHz, D₂O) 177.94, 176.29, 174.71 and 171.74 (4xs, 2x amides, 2x acids), 61.09 and 54.94 (2xd, $2xCH\alpha$), 53.11 (dd, J 37Hz, ¹³CH₂CH), 39.16 (t, ¹³CH₂), 35.23 (t, CH₂CO), 30.26 (t CH_CH₂CH₂), 21.35 (t, $CH_2CH_2CH_2CH$, 19.22 and 17.61 (2xq, MeCD); m/z (+ve argon FAB), MH⁺ (disulphide) 729, MH+ (thiol) 366.

 δ - $(L-\alpha$ -Aminoadipovl)-L- $(3-13C)$ cysteinvl-D-valine disulphide $(3f)$. The tripeptide $(N\text{-}benzyloxycarbonyl)\text{-}α-benzvlester-\delta-(L-\alpha-Aminoadipovl)\text{-}S-benzvl L$ -(3-¹³C)cysteinyl- D -valine benzylester (4f)⁷ (0.10g, 0.1mmol), was deprotected according to general method C, to give the title tripeptide thiol, which was oxidized to the disulphide form by adjusting the pH of the solution to 9-10, and bubbling $O₂$ through for 4h. This was purified by reverse phase HPLC (mobile phase 20mM $NH_4HCO_3/20\%$ MeOH), to give the title compound as a white powder (24mg, 67%); δ_H $(500MHz, D₂O), 4.07$ (1H, d, J 5Hz, CH α of valine), 3.73 (1H, dd, J 2x6Hz, CH α of aminoadipoyl), 3.37-3.23, 3.15-3.04 and 2.87-2.82 (2H, AB part of ABMX system, J 143, 142, 14, 9, 5Hz ¹³CH₂S), 2.40 (2H, dd, J 2x7Hz, CH₂CO), 2.13-2.06 (1H, m, CHMe₂), 1.94-1.81 and 1.78-1.64 (4H, 2xm, $CH_2CH_2CH_2CO$), 0.93 and 0.89 (6H, 2xd, J 7Hz, CHMe₂); δ _C (126MHz, D₂O), 178.11, 176.29, 174.74 and 171.70 (4xs, 2x amides, 2x acids), 61.04 and 54.87 (2xd, $2xCH\alpha$), 53.14 (dd, J 37Hz, ¹³CH₂CH), 39.18 (t, ¹³CH₂), 35.23 (t, $CH₂CO$), 31.10 and 30.26 (t and d, CHCH₂CH₂ and CHMe₂), 21.34 (t, $CH_2CH_2CH_2$), 19.34 and 17.73 (2xq, MeCH); m/z (+ve argon FAB), MH⁺ (disulphide) 727, MH+ (thiol), 365.

δ -(L- α -Aminoadipovl)-L-(3-¹³C)cysteinvl-(¹⁵N)glycine thiol (5b).

The protected tripeptide (6) (O.O89g, 0.12mmol) was deprotected according to general method C. without subsequent oxidation to give after work up the title compound as a gummy solid $(0.023g, 59\%)$; δ_H (500MHz, D₂O, pH 6), 4.56-4.54 (1H, m, X part of ABMX system, CH α of cysteine), 3.79 (2H, d, A₂ part of A₂X system, J 1.5Hz, C_{H₂ of glycine),} 3.75 (1H, dd. J 2x6Hz, CH α of aminoadipoyl), 3.11-3.05 and 2.81-2.77 (2H, AB part of ABMX system, J 144, 144, 14, 7, 5.5Hz, CH₂S), 2.43 (2H, dd, J 2x7Hz, CH₂CO), 1.93-1.87 and 1.77-1.67 (4H, 2xm, CH₂CH₂CH₂CO); δ_C (126MHz, D₂O), 176.24, 173.23 and 172.82 (3xs, 2x acids, 2x amides), 55.83 (d, $\mathbb{C}H\alpha$ of aminoadipoyl), 53.36 (d of d, J 35Hz, $\mathbb{C}H\alpha$ of cysteine), 41.46 (d of t, J 13Hz, $CH₂15NH$), 34.77 (t, CH₂CO), 29.50 (t,

CH₂CH₂CH₂CO), 26.50 (t, ¹³CH₂S), 20.93 (t, CH₂CH₂CH₂); m/z (+ve argon FAB), MH⁺ 324.

Preparation of N-benzyloxycarbonyl-serine aldehyde propylamide (7).

Penicillin G sulphone24 (O.lOg, 0.3mmol) was stirred **in n-propylamine (5ml) at room temperature for** 2h, before removal of the **solvent in vacua, The residue was taken up** in HCl (1M,10ml), saturated with brine (10ml), and extracted with EtOAc (3x15ml). The **combined** organic layers were washed with brine **(saturated solution, 20ml), dried** (Na2S04), filtered, and the solvent removed in **vacua, to give a clear gum** (0.061g. 84%); whose ¹H and ¹³C n.m.r. spectra were observed in different solvents. (a) .

 δ H (200MHz, CDCl₃), 9.49 (1H, s, HCO), 7.47-7.20 (5H, m, ArH), 7.03 (1H, d, J 8Hz, NH), 6.80-6.73 (1H, unresolved t, NHCH₂), 4.98 (1H, d, J 8Hz, CH α), 3.67 (2H, s, CH₂Ar), 3.16 (2H, overlapping 2xt, NHC H_{2}), 1.50-1.46 (2H, m, C H_{2} Me), 0.86 (3H, t, MeCH₂); δ C $(50MHz, CDCl₃), 196.15$ (d, HCO), 172.45 and 163.62 (2xs, 2x amides), 134.28 (s, ArC-1), 129.52-126.99 (3xd, 3xArCH), 63.17 (d, CH α), 42.85 and 41.53 (2xt, CH₂Ar and NHCH₂), 22.28 (t, CH₂Me), 11.03 (q, MeCH₂); m/z (DCI, NH₃) MH⁺ (263, 100%), MNH₄⁺ (280, 15%).

(b) .

 δ H (500MHz, CD3SOCD3), 9.50 (0.5H, s, HCO), 8.21 and 7.73 (0.5H, 2xs, CHOH, (E) and (Z)), 7.30-7.12 (7H, m, ArH and 2xNH), 5.08 (1H, d, CH α), 3.60 and 3.52 (2H, s, CH₂Ar), 3.08-3.05 (2H, 3xm, NHCH₂), 1.49-1.47 (2H, m, CH₂Me), 0.86-0.72 (3H, t, MeCH₂); δ C (126MHz. CD\$S02CD3), 196.15 (d, HCO), 172.15, 171.32, 170.06, 169.93, 165.40, 164.71. 158.83 and 146.66 (6xs, 2xd, 6x amides and $2xC=CHOH$), 136.42 and 136.35 (2xs, $2xArC-1$), 129.59-126.33 (d's, Ar CHs), 110.02 and 106.54 (2xs, $C=CHOH$), 63.38 (d, $CH\alpha$), 42.52, 42.06, 41.99, 41.49, 40.63 and 40.39 (6xt, $3xCH₂Ar$ and $3xNHCH₂$), 22.44, 22.22 and 21.94 (3xt, $3xCH₂Me$), 11.26 and 11.20 (q's, MeCH₂). (data represents mixture of aldehyde (50%) and enols (Z) and (E) (50%) . Irl.

 δ C (126MHz, CD₃SOCD₃ and D₂O (5:1)), 195.69 (d, HCO), 172.02, 171.29, 171.04, 170.67, 169.40, 165.47, 147.33 (6xs. 1xd, 6xamides and $1xC=CHOH$), 135.82 and 135.45 (2xs. $2xArC-1$), 129.48-126.44 (d and s, ArCH and ArC), 109.02 (s, C=CHOH), 88.85 (d, $CH(OH)_2$, 41.94, 41.43, 40.80, 40.25, (4xt, CH_2Ar and $NHCH_2$), 22.20 and 21.94 (2xt, $2xCH₂Me$), 11.22 and 10.04 (q's, $MeCH₂$). (data represents mixture of aldehyde/enol/hydrate).

(d) .

 δ (126MHz, CD₃SOCD₃ and H₂O (5:1)), as for (c) except 59.71 (d, CH α) is observed. (e) .

 δ_C (126MHz, CD₃SOCD₃ and excess D₂O (1:1)), 172.08, 169.50 (2xs, 2x amides), 135.14 $(s, Arc-1), 129.31-126.68$ (d's, Ar $CH(s)$, 88.85 (d, $CH(OH)_2$), 41.83 and 40.38, (2xt, $CH₂Ar$ and NHCH₂), 21.66 (t, 2xCH₂Me), 10.76 (q, $MeCH₂$), (data represents hydrate).

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 $C_{14}H_{18}N_2O_3$ 1/3 H₂O requires C 62.67, H 7.01, N 10.44%, found C 62.67, H 7.41, N 10.77%. The 2,4 dinitrophenylhydrazone derivative was prepared in the usual manner, and recrystallised from water/EtOH to give; δ_H (300MHz, CDCl3/CD3OD), 9.03 $(1H, d, J, 2.6Hz, ArcH-3), 8.20 (1H, dd, J, 9.5, 2.6Hz, ArcH-5), 7.55 (1H, d, J, 4Hz, CH=N),$ 7.50 (1H, d, J 9.5Hz, ArCH-6), 7.35-7.32 (5H, m, ArH), 5.15 (1H, d, J 4Hz, CH α), 3.64-3.56 (2H, m, CH₂Ar), 3.14 (2H, t, J 7Hz, NHCH₂), 1.52-1.45 (2H, tq J 7Hz, CH₂CH₂), 0.87 (3H, t, J 7Hz, MeCH_2); C₂₀H₂₂N₆O₆ requires 54.29, H 5.01, N 19.00%, found C 54.12, H 5.11, N 19.13%.

Preparation of N- $(4$ -methoxybenzyloxycarbonyl)- α - $(4$ - $\text{methoxybenzvlester}-\delta-L-\alpha \text{-aminoadipov}$. L-vinvlglycine (11).

Diprotected aminoadipic acid $[N-(4-methoxybenzyloxycarbonyl)-\alpha-(4-methoxybenzylox/carnony)]$ methoxybenzylester)- δ - \underline{L} - α -aminoadipic acid], (0.20g) and 1 equivalent of triethylamine $(67\mu l)$, were dissolved in freshly distilled THF $(6ml)$, and treated with isobutyl chloroformate (leq, 60μ l) and L-vinylglycine according to general method D. to give upon work up the title compound as a white foam, which was crystallized from Et₂O/Petrol to give the title compound (0.18g, 75%); α_{D} |20 = -6.70 (c=1, acetone); δ_{H} $(500MHz, CDCl₃), 7.29-7.25$ and 6.90-6.84 (8H, $2xA₂B₂$ systems, overlapping $2xA₁$), 6.98 (lH, d, J 7Hz, NH), 5.97-5.89 (lH, 8 line multiplet, CH=CH2), 5.58 (lH, d, J 7Hz, NH), 5.34 (1H, d, J 17Hz, trans-HCH=CH), 5.30 (1H, d, J,10Hz, cis-HCH=CH), 5.10-4.98 (5H, $2xCH_2$ Ar and CH α of Vinylglycine), 4.40 (1H, m, CHCH₂), 3.78 (6H, s, 2xMeO), 2.42-2.25 (2H, m, CH₂CO), 1.96-1.80 and 1.75-1.64 (4H, CH₂CH₂CH₂CO); δ_C (50MHz, CDCl₃), 173.49, 173.07 and 172.64 (3xs, 2x amides, 1x acid, 1x ester), 159.94 and 159.79 (2xs, $2xArC-$ 4,). 156.79 (s, -urethane), 132.05 (d, cH=CH2), 130.31-127.46 (d and s, ArCH, and Arc-1), 118.05 (t, $CH_2=CH$), 113.99 (2xd, 2xAr $C=3$), 67.10 and 66.99 (2xt, OCH_2Ar), 55.18 $(2xq, 2xMe)$, 54.47 and 53.46 $(2xd, 2xCH\alpha)$, 35.01 (t, CH₂CO), 31.34 (t, CH₂CH), 21.16 (t, CH_2CH_2CO ; m/z (+ve argon FAB), MNa+ (551), MH+ (529).

Preparation of N-(4-methoxybenzvloxycarbonyl)- α -(4 m ethoxybenzylester) - δ -L - α -aminoadipoyl - L - vinylglycine - D - valine benzhvdrylester (10a).

The dipeptide N-(4-methoxybenzyloxycarbonyl)- α -(4-methoxybenzylester)- δ - \underline{L} - α aminoadipoyl- L -vinylglycine (LL -AVinylglycine) (11), (0.245g, 0.45mmol) was coupled to the free amine of the benzhydryl ester of D -Valine (1.1eq) with EEDQ, according to general method A, to give after purtfication by column chromatography on silica gel [eluant DCM/EtOAc $(2:1)$ Rf 0.6] the title compound as a white foam $(0.28g, 77\%)$; δ H $(500MHz, CDCl₃)$, 7.36-7.24 and 6.90-6.84 (19H, 2xm, ArH and CHPh₂), 6.74 (lH, d. J 9Hz, NH), 6.49 (lH, d, J 8Hz, NH), 5.89-5.83 (lH, complex m, 8 lines, _TT _TT-\ Z rn ,,-rl > r OII- *TrT\ Z ,?,Y 1-11 I **1 q-l.1 Lfl=LrlzJ, 3.3u (Ill. cl, J ix-lz, 1Ul-l). 3.33 (lrl,** a, **J I/HZ, irans-HCii=Ciij, 5.25** (iH, d, J 10Hz, ξ is-HCH=CH), 5.08-4.97 (5H, m, 2xOCH₂Ar and CH α), 4.65 (1H, 4 lines, X part of ABX system, J 2x9Hz, CHα), 4.35-4.34 (1H, m, CHα), 3.78 and 3.76 (9H, s, 3xMeO), 2.25-2.12 (3H, 2xm, CH₂CO and CH_B of valine), 1.81-1.64 (4H, 2xm, CH₂CH₂CH₂CO),

0.87 and 0.76 (6H, 2xd, J 7Hz, Me₂CH); δ _C (126 MHz, CDCl₃), 172.13, 170.71, 169.75, (4xs, 2x esters, 2x amides), 159.76, 159.66 (2xs, 2x Ar_C-4), 156.12 (s, OCONH), 139.53-139.36 (2xs, Ph_C-1), 133.74 (d, C_H=CH₂), 130.11-126.96 (d and s, ArCH and Ar C -1), 118.52 (t, $CH_2=CH$), 114.02-113.91 (2xd, Ar C -3), 78.14 (d, $CHPh_2$), 66.97 and 66.80 (2xt, CH₂Ar), 57.20 and 55.63 (2xd, 2xCH α), 55.21 (3xq, 3xMeO), 53.52 (d, CH α), 35.32 (t, CH₂CO), 31.80 (t, CHCH₂CH₂), 31.35 (d, CHMe₂), 21.33 (t, CH_2CH_2CO), 19.02 and 17.25 (2xq, 2xMe); m/z (+ve argon FAB), MH⁺ (794); $C_{45}H_{51}N_3O_{10}$ requires C 68.08, H 6.47, N 5.29% : found C 68.34, H 6.77, N 5.01%; m.p. $97-103$ °C.

Preparation of N- $(4$ -methoxybenzyloxycarbonyl) $-\alpha$ - $(4$ methoxybenzylester) - δ - L - α - aminoadipoyl - L - vinylglycine - glyci benzhydrylester (10b).

The title compound was prepared in an analogous manner to the valinyl analogue, with LL-AVinylglycine (0.27g, 0.55mmol), EEDQ, and glycine benzhydryl ester, to give after purification by column chromatography on silica gel [eluant DCM/EtOAc (1:1) Rf 0.55] the title compound as a white solid, $(0.21g, 55\%)$; m.p 115-119°C; δ_H (SOOMHz, CDC13), 7.35-7.26 and 6.91-6.86 (19H, 2xm, ArH and CEPhz), 6.58 (lH, d, J 5Hz, NH), 6.41 (lH, d, J 7Hz, NH), 5.88-5.81 (lH, complex m, 8 lines CH=CHz), 5.47 (1H, d, J 8Hz, NH), 5.38 and 5.28 (2H, 4 lines, J 17Hz, 10Hz, CH₂=CH), 5.47-5.09 (5H, m, $2xCH_2O$ and CH α of vinylglycine), 4.35-4.34 (1H, m, CH α), 4.19-4.09 (2H, AB part of ABX system, C H_2 of glycine), 3.79 (6H, s, 2xMeO), 2.31-2.13 (2H, m, C H_2 CO), 1.90-1.75 and 1.72-1.60 (4H, 2xm, CH₂CH₂CH₂CO); δ_C (50MHz, CDC₁₃), 172.53, 170.49, 169.03 (4xs, 2x amides, 2x esters), 159.96 and 159.75 (2xs, 2xArC-4), 156.44 (s, OCONH), 139.61 (s, PhC-1), 133.40 (d, CH=CH₂), 130.31-127.05 (d and s, ArCH and Ar C -1), 118.52 (t, $CH_2=CH$), 114.06-113.97 (2xd, Ar C -3), 78.14 (d, $CHPh_2$), 66.99 and 66.78 (2xt, CH₂Ar), 55.60 (d, CH α), 55.19 (3xs, 3xMeO), 53.52 (d, CH α), 41.51 (t, CH_2 of glycine), 35.10 (t, CH₂CO), 31.52 (t, CH₂CH₂CH₂CO), 21.12 (t, CH₂CH₂CH₂); $C_{42}H_{45}N_3O_{10}$ requires C 67.10, H, 6.03, N 5.59%, found C 66.84, H 6.10, N 5.27%; m/z (+ve argon FAB), MH+ 752.

$Preparation of δ -L- α -aminoadipovl-L-vinvleveine-D-valueline (9a).$

The protected tripeptide $(10a)$ $(0.086g, 0.11mmol)$ was deblocked with TFA/Anisole (5:l) according to general procedure B, to give after purification by reverse phase HPLC (mobile phase $20mM NH₄HCO₃/20%$ MeOH), the title compound (0.028g, 77%); δ H (500MHz, D₂O, pH 6), 5.96-5.89 (1H, 8 lines, CH=CH₂), 5.42 (1H, d, J 17Hz, trans-HCH=CH), 5.38 (1H, d, J 10Hz, cis-HCH=CH), 4.92 (1H, d, J 7Hz, CH α of vinylglycine), 4.72 (1H, d, J 6Hz, $CH\alpha$ of valine), 3.72 (1H, t, J 5Hz, $CH\alpha$ of L- α -AA), 2.38 (2H, dd, J 2x7Hz, CH₂CO), 2.20-2.09 (1H, m, CHMe₂), 1.94-1.63 (4H, 2xm, CH₂CH₂CH₂CO), 0.89 and 0.87 (6H, 2xd, J 5Hz, Me₂CH); δ_C (126MHz, D₂O), 177.22, 175.84, 174.60, and 172.16 (4xs, 2x acids, 2x amides), 132.31 (d, CH=CH2), 119.96 (t, CH= CH_2), 60.17, 56.88 and 54.81 (3xd, 3x $CH\alpha$), 35.11 (t, CH₂CO), 31.01 (t, $CH_2CH_2CH_2CO$), 30.23 (d, CHMe₂), 21.28 (t, CH₂CH₂CH₂CO), 19.13 and 17.57 (2xq, $Me₂CH$; m/z (+ve argon FAB), MH⁺ 344.

Preparation of δ **-L-** α **-aminoadipovl-L-vinylglycine-glycine (9b).** This was prepared in an analogous manner to $(9a)$ from the tripeptide $(10b)$ (O.OSSg, O.l2mmol), to give after purification by reverse phase HPLC (mobile phase

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 20mM NH_4 HCO₃), the title compound (0.031g, 91%); δ_H (500MHz, D₂O, pH 6), 5.97-5.91 (1H, 8 lines, CH=CH₂), 5.42 (1H, d, J 17Hz, trans-HCH=CH), 5.39 (1H, d, J 10Hz, cis-HCH=CH), 4.92 (1H, d, J 6Hz, CH α of vinylglycine), 3.79-3.70 (3H, m, AB system of glycine CH₂ and CH α of L- α -AA), 2.39 (2H, dd, J 2x7Hz, CH₂CO), 1.90-1.83 and 1.74-1.65 (4H, 2xm, CH₂CH₂CH₂CO); δ _C (126MHz, D₂O), 176.53, 176.03, 174.79, and 172.16 (4xs, 2x acids, 2x amides), 132.20 (d, $CH=CH_2$), 120.00 (t, $CH=CH_2$), 56.65 and 54.90 (2xd, 2x $\text{CH}\alpha$), 43.86 (t, $\text{CH}\alpha$) of glycine), 35.21 (t, $\text{CH}\alpha$ ₂CO), 30.25 (t, CH_2CH_2CHO , 21.30 (t, $CH_2CH_2CH_2CO$); m/z (+ve argon FAB), MH⁺ 302, MNa⁺ 324.

<u>ylester-δ-L-α-aminoadipoyl-O-benzyl-D-</u> werth *weight benzylester* (14b)

N-Benzyloxycarbonyl-a-benzylester-8-L-a-Aminoadipoyl-O-benzyl-D-Serine, was prepared according to general method D from dibenzyl protected δ -L- α aminoadipic acid $(0.29g)^7$, and O-benzyl- D -serine $(0.15g)$, to give a colourless gum $(0.34g\ 78\%)$; $\delta H (200MHz, CDCl₃)$, 9.30 (1H, vbr s, HOOC), 7.43-7.26 (15H, complex m, ArH), 6.79 (lH, d, J 6Hz, NH), 5.82 (lH, d, J 7Hz, NH), 5.21-5.05 (4H, 2xABq, $2xA₁CH₂O₂C-1$, 4.78-4.47 (1H, m, CH α of aminoadipoyl), 4.44 (2H, s, OCH₂Ar), 4.42-4.38 (1H, m, CH α of serine), 3.81-3.75 and 3.71-3.65 (2H, AB part of ABX system, CH₂O), 2.25-2.14 (2H, m, CH₂CO), 1.96-1.61 (4H, 2xm, CH₂CH₂CH₂CO). This material (0.34g, 0.6mmol) was coupled to \underline{D} -valine benzyl ester, tosylate salt (0.23g, 0.6mmol), according to general method A, and the crude material was purified by column chromatography on silica gel [eluant DCM/EtOAc (9:l) Rf 0.41, to give the title compound as an oil (0.23g, 75%), $[\alpha_{D}]^{20}$ = -7.5° (c=1.06, DCM); δ_{H} (500MHz, CDC13). 7.41-7.28 (20H, m, ArH), 7.13 (lH, d, 3 8Hz, NH), 6.39 (lH, d, J 6Hz, NH), 5.52 (1H, d, J 8Hz, NH), 5.21-5.05 (6H, m, 3xCH₂Ar), 4.62-4.37 (5H, m, 3xCH α and CH₂Ar), $3.91-3.86$ and $3.51-3.45$ (2H, AB part of ABX system J 9, 8, 4Hz, CH₂OCH₂Ar), 2.29-2.10 (2H, m, CH₂CO), 1.91-1.80 (3H, m, CHMe₂ and CH₂CH₂CH₂CO), 1.69-1.65 (2H, m, CH₂CH₂CO), 0.87 and 0.77 (6H, 2xd, J 7Hz, Me₂CH); δ _C (50MHz, CDCl₃), 172.63, 172.42 , 171.69 and 170.53 $(4xs, 2x \text{ esters}, 2x \text{ amides}), 156.29$ $(s, \text{ urethane}),$ 137.40, 136.39 and 135.44 (3xs, 3xArC-1), 128.76-128.12 (d's, ArC-H), 73.51, 69.45, 67.15, 67.06 and 66.96 (5xt, $4xArCH_2$ and $CHCH_2OCH_2$), 57.36, 53.65 and 52.17 (3xd, 3xCH α), 35.29 (t, CH₂CO), 31.63 (t, CH₂CH₂CH₂), 30.90 (d, CHMe₂), 21.07 (t, $CH_2CH_2CH_2$), 18.86 and 17.27 (2xq, 2x Me_2CH); m/z (+ve argon FAB), MH⁺ 752;.oil.

$N-Benzvloxvcarbonvl-\alpha-benzvl-\delta-L-\alpha-aminoadipoyl-O-benzyl-L-serinyl-$ D-valine, benzylester (14a).

This tripeptide was prepared in an analogous manner to its diastereoisomer, to give after purification by column chromatography on silica gel [eluant DCM/EtOAc (9:1)], the title compound as a white solid, which was recystallized from DCM/Petrol, (0.31g, 89%); 8~ (SOOMHz, CDC13), 7.54-7.27 (20H, m, ArH), 6.88 (lH, d, J 7Hz, NH), 6.39 (1H, d, J 7Hz, NH), 5.58 (1H, d, J 8Hz, NH), 5.16-5.04 (6H, m, $3xCH₂Ar$), 4.65-4.36 (5H, m, $3xCH\alpha$ and CH₂Ar), 3.91-3.84 and 3.52-3.45 (2H, AB part of ABX system, J 7, 4, 4Hz, CH₂OCH₂Ar), 2.23-2.11 (2H, m, CH₂CO), 1.90-1.80 (1H, br m, CHMe₂), 1.69-1.49 (4H, 2xm, CH₂CH₂CH₂CO), 0.87 and 0.76 (6H, 2xd, J 7Hz, <u>Me</u>₂CH);

m/z (+ve argon FAB) MH⁺ 752; $[\alpha_{D}]^{20} = 1.4^{\circ}$ (c=1.05, DCM); C₄₃H₄₉N₃O₉ requires, C 68.69, H 6.57, N 5.59%, found C 68.64, H 6.50, N 5.09%; m.p. 113-118'C.

.
(L-α₂Aminoadipoyl)-L-serinyl-D-valine (13a^t)

N-Benzyloxycarbonyl-a-benzylester-6-l-a-aminoadipoyl-O-benzyl-L-serinyl-Dvaline, benzylester (14a) (O.lg, O.l3mmol), was dissolved in THF (8ml) and water (8ml), and Pd/C (lo%, 20mg) was added. The system was evacuated and flushed with hydrogen gas $(x3)$ before stirring overnight at 40° C. The solution was then filtered through washed celite, and the THF removed in vacuo. The aqueous layer was washed with DCM (30ml), and EtOAc (30ml) before freeze drying. The material recovered was then purified to homogeneity by reverse phase HPLC (mobile phase 25mM NH₄HCO₃), to give the title compound as a white powder in quantitative yield. δ_H (500MHz, D₂O), 4.30 (1H, dd, J 2x5Hz, C $\underline{H}\alpha$ of serine), 3.92 (1H, d, J 6Hz, CH α of valine), 3.67 (2H, d, J 5Hz, CH₂O), 3.56 (1H, t, J 5Hz, CH α of aminoadipoyl), 2.25 (2H, dd, J 2x7Hz, CH₂CO), 1.94-1.91 (1H, m, CHMe₂), 1.73-1.63 and 1.56-1.45 (4H, 2xm, CH₂CH₂CH₂CO), 0.72 and 0.68 (6H, 2xd, J 7Hz, Me₂CH); δ_C (126MHz, D₂O), 178.41, 176.47, 174.76 and 171.53 (4xs, 2x acids, 2x amides), 61.65 (t, CH₂OH), 60.87, 56.11 and 54.77 (3xd, $3xCH\alpha$), 35.09 (t, CH₂CO), 31.00 (d, CHMe₂), 30.16 (t, $CH_2CH_2CH_2CO$, 21.19 (t, $CH_2CH_2CH_2$), 19.72 and 17.49 (2xq, 2x Me_2CH); m/z (+ve argon FAB), MH+ 348.

\cdot (L- α -Aminoadipoyl) D-serinyl-D-valine (13b)

This title compound was prepared in an analogous manner to its diastereoisomer (13a) from (14b), to give the title compound as a gummy solid (35mg, 84%); δ ^H (500MHz, D₂O), 4.31 (1H, dd, J 2x6Hz, CH α of serine), 3.93 (1H, d, J 6Hz, CH α of valine), 3.70-3.63 (2H, 8 lines, AB part of ABX system, J 11, 7, 5Hz, $C_{12}O$), 3.56 (1H, dd, J 2x5Hz, CH α of aminoadipoyl), 2.23 (2H, dd, J 2x7Hz, CH₂CO), 1.93 (1H, m, CHMe₂), 1.73-1.67 and 1.56-1.48 (4H, 2xm, CH₂CH₂CH₂CO), 0.73 and 0.71 (6H, 2xd, J 7Hz, Me₂CH); δ _C (126MHz, D₂O), 178.06, 176.49, 174.77 and 171.70 (4xs, 2x acids, 2x amides), 61.49 (t, CH₂OH), 60.76, 56.04 and 54.88 (3xd, 3xCH α), 35.15 (t, CH₂CO), 30.91 (d, CHMe₂), 30.22 (t, CH₂CH₂CH₂CO), 21.34 (t, CH₂CH₂CH₂), 19.01 and 17.59 (2xq, 2x $Me₂CH$); m/z (+ve argon FAB), MH⁺ 348.</u>

Data for ozonolysis products.

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(L.α-Aminoadipoyl) DL [2-amino 1-carboxyl propan 3 - all D valin $(8a)$.

The tripeptide δ -(L- α -aminoadipoyl)-L₂-vinylglycine- D -valine (0.02g, 0.05mmol) (9a), was dissolved in MeOH/H₂O (3:1), (15ml), and cooled to -50 \degree C. Ozone was bubbled through the solution until the development of a pink/blue colouration, and continued through the solution for 30min. Oxygen was bubbled through the solution for 15min, followed by $Ar(g)$ for 1h at -50°C. Dimethylsulphide (1ml) was added, and $Ar(g)$ bubbled through the solution for a further 4h, during which time the solution was warmed to room temperature. Water (20ml) was added, before

removal of the solvent in vacuo, then further water (5ml) was added before freeze drying the resulting solution. This crude material was purified by reverse phase HPLC (stationary phase μ -bondpack, mobile phase 75% (0.05% HCOOH(aq))/25% MeOH) or (stationary phase ODS, mobile phase $10mM$ NH₄HCO₃), to give the title compound as a white solid and as an epimeric mixture of hydrates; δ _H (500MHz, D₂O, pH 4), 5.26 and 5.25 (1H, 2xs, CH(OH)₂), 4.13 and 4.11 (1H, 2xd, CH α of valinyl), 3.72 and 3.70 (1H, 2xt, CH α of aminoadipoyl), 2.41-2.37 (2H, m, CH₂CO), 2.12-2.07 $(1H, m, CHMe_2)$, 1.88-1.64 (4H, 2xm, CH₂CH₂CH₂CO), 0.94-0.89 (6H, m, Me₂CH); δ_C $(126 MHz, D_2O), 177.00-169.50$ (4xs, 2x acids, 2x amides), 89.14 and 89.06 (2xd, $CH(OH)_2$, 59.63 and 54.81 (2xbr d, $CH\alpha$ of aminoadipoyl, and $CH\alpha$ of valinyl), 35.13 (t, CH_2CO) , 30.83 and 30.74 (2xt, CH₂CH₂CH₂CO), 30.14 and 30.09 (2xd, CHMe₂), 21.33 and 21.24 (2xt, $CH_2CH_2CH_2$), 18.96 and 18.91 (2xq, Me_2CH), 17.56 (2xq, Me_2 CH), δ _C (126MHz, H₂O/10% D₂O) data as for above except new signal at 58.59); m/z (+ve argon FAB), (ex (COOH)₂/ HOCH₂CHOHCH₂OH) MH⁺ (hydrate) 364, MH⁺ (enol/aldehyde) 346. δ_H (500MHz, D₂O, pH > 8.5), 8.11 and 7.72 (1H, 2xs, enolic H (E) and (Z)), 4.03 and 3.97 (1H, 2xd, CH α of valinyl), 3.10-3.00 (1H, br m, CH α of aminoadipoyl), 2.27-2.21 and 2.17-2.04 (2H, 2xm, CH₂CO), 2.02-1.84 (1H, 2xm, CHMe₂), 1.55-1.30 (4H, br m, CH₂CH₂CH₂CO), 0.78-0.62 (6H, m (2xd)x2, Me₂CH); δ _C (126MHz, D20, pH>8.5), 177.00-169.50 (4xs, 2x acids, 2x amides), 169.4 and 168.9 (2xd, enolic (E) and (Z) C= $\mathbb{C}H(OH)$), 60.00 and 56.2 (2xbr d, $\mathbb{C}H\alpha$ of aminoadipoyl, and $CH\alpha$ of valinyl both doubled up), 35.9 and 35.8 (t, CH_2CO), 34.2 and 34.1 (2xt, $CH_2CH_2CH_2CO$, 30.1 (2xd, $CHMe_2$), 22.2 (2xt, $CH_2CH_2CH_2$), 19.4 and 19.5 (2xq, $Me₂CH$, 17.8 (2xq, $Me₂CH$).</u></u>

δ -(L- α -Aminoadipovl)-DL-[2-amino1-carboxvl-propan-3-all-glycine $(8b)$.

The tripeptide δ -(L- α -aminoadipoyl)-L-vinylglycine-glycine (0.02g, 0.06mmol) (9b) was ozonolysed in an analogous manner to (9a), to give after purification by reverse phase HPLC the title compound as a powdery solid; δ_H (500MHz, D₂O, pH 4), 5.22 (1H, very tight 2xs, $\Delta \delta$ 2Hz (pH dependant), CH(OH)₂), 3.81-3.76 (2H, d, CH₂ of glycine), 3.67-3.64 (1H, m, CH α of aminoadipoyl), 2.36-2.33 (2H, m, CH $_2$ CO), 1.82-1.77 and 1.66-1.59 (4H, 2xm, CH₂CH₂CH₂CO); δ_C (126MHz, D₂O), 176.30-171.73 (4xbr s, 2x acids, 2x amides), 89.15 (1xd, $CH(OH)_2$), 54.99 (br d, $CH\alpha$ of aminoadipoyl), 43.27 (t, CH_2 of glycine) 35.15 and 34.83 (2xt, CH_2CO), 30.22 (2xt, $CH_2CH_2CH_2CO$), 21.31 and 21.21 (2xt, $CH_2CH_2CH_2$).

. **Qbservatlon** of hvdrate resonances.

The tripeptides (3a-f, 5b) were incubated with IPNS under the standard incubation conditions (method (E)), to give the crude incubation mixtures. These were dissolved in D₂O (400 μ I) and examined by both ¹H and ¹³C n.m.r (at 500 and 126MHz respectively), typically collecting $45,000$ transients for each overnight $13C$ experiment. (Spectral differences were observed for tripeptides (3b),(3e), and gave; δ C (126MHz, D₂O, pH 7-8), 89.10 and 89.14 (2xd, ¹³CH(OH)₂, $\Delta \delta$ 5-11Hz (pH dependent), J_{CH} 167Hz), DEPT analysis indicated the absence of protons 2 or 3 bonds distant to this centre). [Tripeptide (5b) treated in an identical manner gave δ C $(126MHz, D_2O)$ 89.14 (d, ¹³CH(OH)₂).]

Enzyme used for denatured controls was prepared by heat treating $(100^{\circ}C)$ for lmin), and the mixtures were treated identically to above, no signals in the region of the C-13 spectra of interest were observed. Purification of the crude incubation mixtures from the tripeptides (3a,b,e,5b) was accomplished by reverse phase HPLC (a) stationary phase ODS, mobile phase 20mM NH₄HCO₃, or (b) stationary phase μ bondpack-NH2, mobile phase 75% (0.05% HCOOH)/25% MeOH, and treated as indicated below.

Reduction of Hydrate (8c)

The tripeptide (3e) (1Omg) was incubated with IPNS under standard incubation conditions. The residue after protein precipitation and freeze-drying was purified by reverse phase HPLC (system (a)) and the fraction that contained the resonances at δ _C 89 coincided with that of the synthetic material. This was dissolved in water (1ml) and treated with NaBH₄ (ca 200μ g). This mixture was stirred for 3h, before quenching, freeze-drying, and examination by C-13 n.m.r. $(126MHz, D_2O, 45,000$ transients). Examination of the C-13 spectra showed 2 resonances at δ_C 62 (2xt, $CH₂O$, $\Delta\delta$ 20Hz). [The proton spectra before reduction had corresponding signals at δ H 5.25 (d, J_{AX} 167Hz, J Hz ¹³CH(OH)₂, and $\Delta\delta$ _H 3Hz). Reduction of a synthetic sample of the hydrate (8a) was carried out in an analogous manner, purified by HPLC (as above), and gave n.m.r. and m/z data identical to that obtained from a 1:l mixture of the two synthetic tripeptides (13a,b).

<u>g experiment</u>

Tripeptide (3a) (15mg) was incubated with IPNS according to general method E. The crude incubation mixture was examined by ${}^{1}H$ n.m.r. (500MHz); signals assigned to the hydrate resonance (-CH(OH)2); δ_H (500MHz, D₂O) 5.22 (2xs, $\Delta \delta$ 3Hz, CH(OH)2) increased in intensity when a small portion of the sample was doped with synthetic material (8a). The remaining material was purified by reverse phase HPLC (system (a) or (b)), isolating a fraction that had identical retention times to that of the synthetic sample. Examination of this fraction by ${}^{1}H$ n.m.r. spectroscopy (3000) transients) clearly showed resonances associated with the hydrate resonances. These were enhanced when doped with synthetic material (8a).

Investigation of Hydrate stability,

The hydrated aldehyde (8a or 8b) was incubated under normal conditions (general method E), followed by the usual work up. H-l n.m.r. showed the presence of the intact hydrate, indicating its stability to incubation conditions.

Preparation and alkylation of ene-thiol (17).

The azetidinone $(15a)^{12}(8mg, 0.013mmol)$ was dissolved in dilute HCl (1ml, pH 2) at room temperature before bubbling H2S through the solution for 45s. The resulting black precipitate was immediately removed by filtration and the colourless solution of mercaptoazetidinone was degassed under high vacuum for 90s in order to remove excess H_2S . The thiol was then rapidly transfered to a solution containing the co-factors required by IPNS (19ml $25mM$ NH₄HCO₃, DTT (7.7mg), ascorbic acid (4.4mg), and $FeSO₄·7H₂O$ (1.4mg). After allowing the solution to stand at room

temperature in the presence of O_2 for 1h the decomposition mixture was freezedried. The crude solid (containing (17)) was re-dissolved in D₂O, prior to H-1 and overnight C-13 n.m.r. spectra being obtained. δ H (500MHz, D₂O), 8.11 (1H, s, C=CHSH), 4.14 (1H, d, J 6Hz, CH α of valinyl), 3.70 (1H, dd, J 6.5, 6Hz, CH α of aminoadipoyl), 2.43-2.41 (2H, m, CH₂CO), 2.14-2.10 (1H, m, CHMe₂), 1.96-1.71 (4H, 2xm, CH₂CH₂CH₂CO), 0.98-0.88 (6H, m, CHMe₂); δ_C (126MHz, D₂O), 179.19, 175.91, 175.98 and 166.47 (4xs, 2x amides, 2x acids), 151.50 (d, C= $CHSH$), 125.66 (s, C=CSH), 60.80 (d, \angle H α of valinyl), 55.05 (d, \angle H α of aminoadipoyl), 35.39 (t, \angle H α CO), 31.61 (t, CH_2CH_2CH , 30.61 (d, $CHMe_2$), 21.61 (t, $CH_2CH_2CH_2CO$), 19.31 and 17.70 (2xq, 2xMe). The solution of ene-thiol was treated with a large excess of iodoacetic acid (15eq) and the pH adjusted to 7.8 before allowing to stand overnight. The crude mixture was freeze-dried prior to purifying the alkylated ene-thiol using reverse phase HPLC (ODS, mobile phase 87% (0.05% HCOOH(aq))/13% MeOH, to give for (18); δ_H (500MHz, D₂O), 7.45 (1H, s, C=CHSCH₂), 4.17 (1H, d, J 6Hz, CH α of valinyl), 3.75 (1H, dd, J 2x6Hz, CH α of aminoadipoyl), 3.61 (2H, s, SCH₂COO), 2.50-2.46 (2H, m, CH₂CO), 2.12-1.97 (1H, m, CHMe₂), 1.94-1.72 (4H, 2xm, CH₂CH₂CH₂CO), 0.89 and 0.84 (6H, m, CHMe₂); δ _C (126MHz, D₂O), 178.60, 176.51, 175.93, 174.95 and 164.47 (5xs, 2x amides, 3x acids), 139.26 (d, C= $CHSCH_2$), 123.65 (s, C=CHSH), 61.24 (d, CH α of valinyl), 55.00 (d, $CH\alpha$ of aminoadipoyl), 38.61 (t, SCH_2COO), 35.28 (t, CH_2CO), 31.44 (d, $CHMe_2$), 30.52 (t, $CH_2CH_2CH_2CO$) 21.61 (t, $CH_2CH_2CH_2CO$), 19.31 and 17.71 (2xq, $2xMe$; m/z (+ve argon FAB) MNa⁺ 442, MNa₂⁺ 464, MNa₃⁺ 486.

Preparation of Thiazole (16b)¹³

Tri(PNB)-protected thiazole (16a)25 (7mg. O.O09mmol), was dissolved in MeOH (1ml), before adding HCl $(1M, 0.25m)$, and heating at reflux for 2h. After cooling the reaction mixture was concentrated in vacuo and the residual acidic solution was neutralized with NaHCO₃ (saturated solution), and extracted with DCM (10ml). After drying $(Na₂SO₄)$, the organic phase was concentrated and a portion of this crude product was hydrogenated overnight at room temperature (latm), in THF/Water $(1:1)$, with $(Pd/C (10%) 2.5mg)$. The reaction was then filtered through a celite pad before concentrating, and purification of the thiazole (16) by reverse phase HPLC (mobile phase 40% (0.05% HCOOH(aq))/60% MeOH); δ H (500MHz, D₂O), 7.95 (1H, s, C=CHS), 4.19 (1H, d, J 6Hz, CH α of valinyl), 3.61 (1H, dd, J 6, 5Hz, CH α of aminoadipoyl), 2.96 (2H, dd, J 2x7Hz, CH₂CNS), 2.12-2.04 (1H, m, CHMe₂), 1.82-1.69 (4H, 2xm, CH₂CH₂CH₂CN), 0.84 and 0.82 (6H, dd, J 2x7Hz, CHMe₂); δ _C (126MHz, D₂O), 177.5, 175.0, 173.5 (3xs 2x amides, 2x acids), 163.5 (s, thiazole C-2), 148.0 (s, thiazole C-4), 126.0 (d, thiazole C-5), 61.0 (d, $\mathcal{L}H\alpha$ of valinyl), 55.3 (d, $\mathcal{L}H\alpha$ of aminoadipoyl), 33.5 (t, CH_2CO), 31.6 (t, $CH_2CH_2CH_2CO$), 30.5 (d, CHMe₂), 25.8 (t, $CH_2CH_2CH_2CO$, 19.5 and 18.1 (2xq, 2x Me); m/z (+ve argon FAB) MH⁺ 344.</u>

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- 25. Prepared from $tri(PNB)$ -protected isopenicillin N by the methods described in reference 12.

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